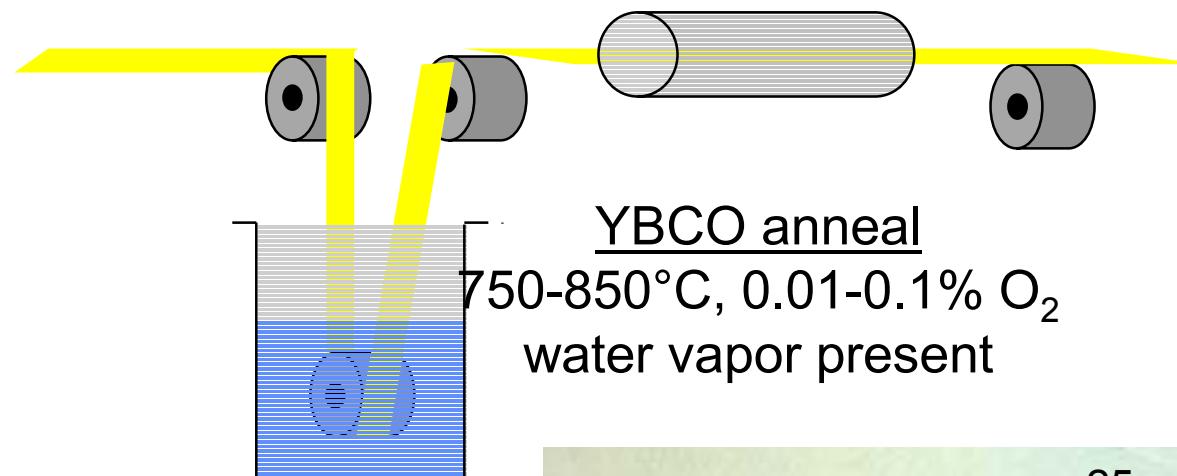
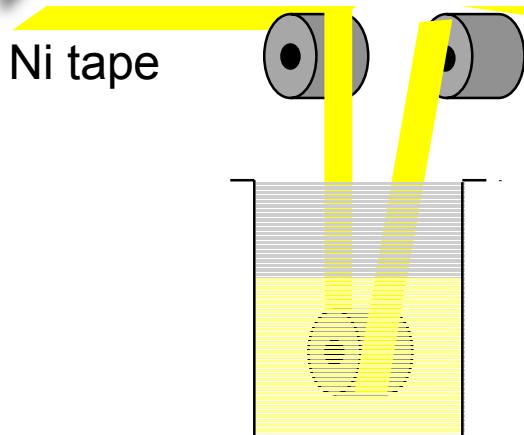


Sandia All-Solution Deposition Method

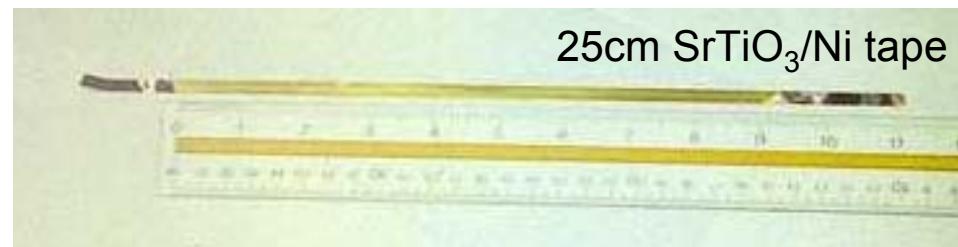
J. Voigt*, P. Clem, M. Siegal, J. Dawley



Solutions are:
metal acetates
metal alkoxides
trifluoroacetic acid
solvents

Coating parameters:
1-5 cm/s pull speed
0.1-0.4 $\mu\text{m}/\text{coating}$
 $3 \text{ cm/s} = 947 \text{ km/yr}$
(single width)

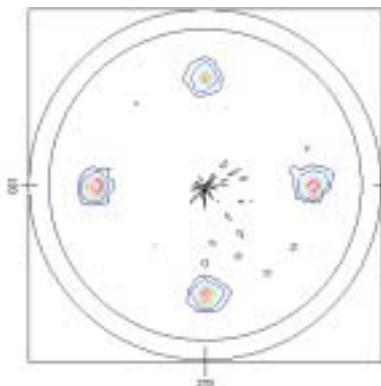
60 sec hot zone = 1.8 m



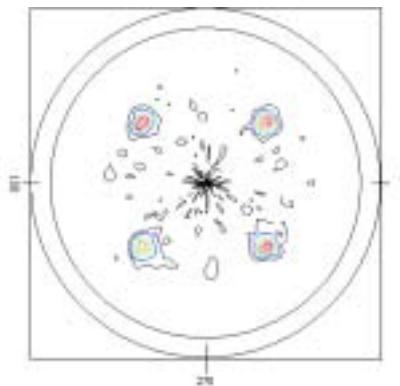


Current status

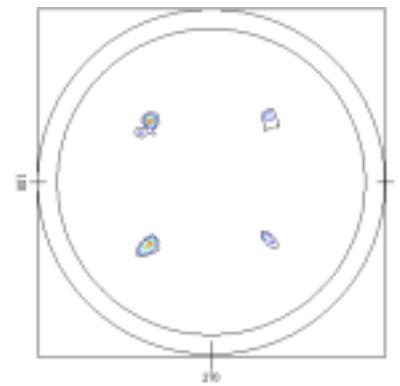
YBCO (113)



STO (220)



Ni (220)



short length (2.5-25 cm)

<u>YBCO thickness</u>	<u>Jc</u>	<u>Ic</u>	<u>substrate</u>
0.25 um	4 MA/cm ²	100 A/cm	LaAlO ₃
1.3 um (6 layers)	0.7 MA/cm ²	91 A/cm	LaAlO ₃
0.12um	1.3 MA/cm ²	16 A/cm	STO/Ni
0.25, 0.5, 1.0 um	- <i>in progress FY2003</i> -		STO/Ni

long length (1 m)

- *in progress FY 2003* -



Scale-up considerations

- Both STO and YBCO may be coated at $3 \text{ cm/s} = 950 \text{ km/yr}$
- Very low capital cost (conductor material cost $\sim \$ 0.39\text{-}0.78/\text{kA}\cdot\text{m}$)
- Limitation of TFA-YBCO: thick film conversion, time

dip coating → pyrolysis → crystallization → oxygenation

3-10 cm/s	15s (was 3h)	0.5-1 h	0.5 h
<hr/>			
1-1.5 hours			

- Approaches:
 - Spontaneous pyrolysis: organic burnout (was 3+ hours, now 20 sec)
 - Thicker YBCO layers: increased to $0.6\text{-}1.0 \mu\text{m}/\text{coating}$
 - Thick film conversion by PO_2 , PH_2O changes (with ORNL)
 - Cima vacuum TFA-YBCO conversion (2 mins. for 0.9 micron film)



Obstacles to scale-up

Thickness of YBCO layers: single coat to 1 micron needed

- Need to control film tensile stress to avoid cracking
- If limited to $< 1\mu\text{m}$, can (i) multi-pyrolyze or (ii) multi-crystallize
 - Demonstrated limited success using both methods

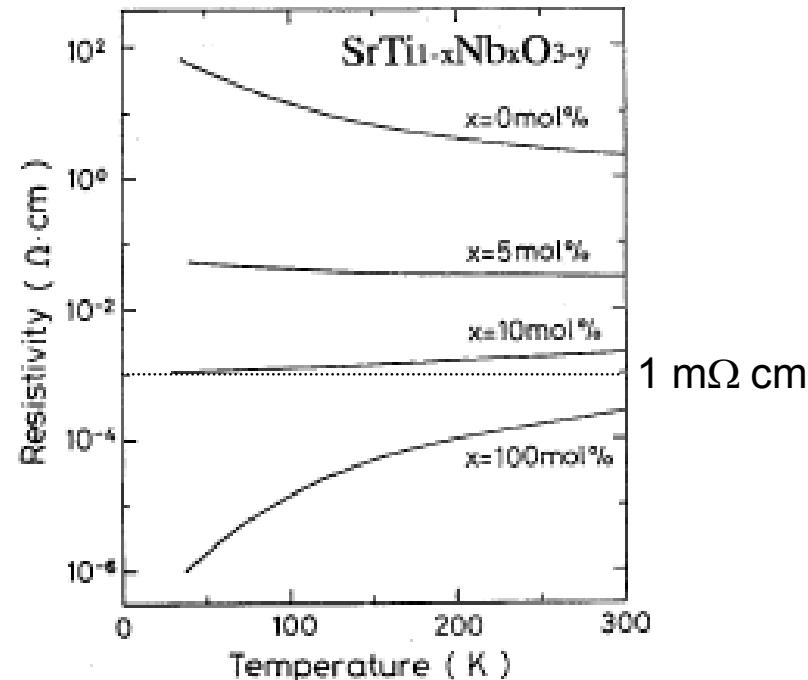
Thick film conversion should/must be faster than 1 hour

- Same issues as BaF_2 -YBCO conversion

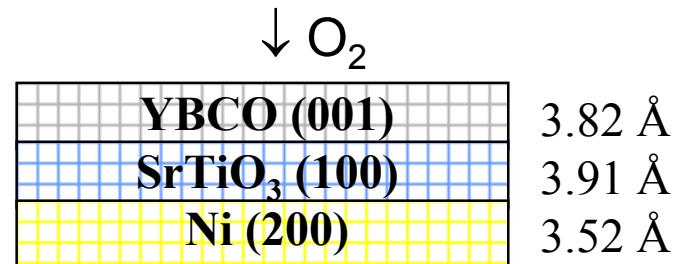
Ability to single-coat buffer layer, with rapid thermal annealing

Why SrTiO₃?

- Compatible with YBCO
- Can be n-type doped to $\rho = 1000 \mu\Omega\text{cm}$, *with* low O₂ diffusion rates
- Low theoretical oxygen diffusion rate



<u>material</u>	<u>unit cell</u>	$D_{\text{ox}}(800^\circ\text{C})$	<u>O diffusion coefficient</u>
MgO	4.20 Å	$2 \times 10^{-19} \text{ cm}^2/\text{s}$	
CaTiO ₃	3.82 Å (orth.)	$1 \times 10^{-15} \text{ cm}^2/\text{s}$	
SrTiO ₃	3.91 Å	$2 \times 10^{-12} \text{ cm}^2/\text{s}$	
LaSrMnO ₃	3.90 Å		
Ho ₂ O ₃	3.75 Å	$7 \times 10^{-7} \text{ cm}^2/\text{s}$	
YSZ	3.63 Å	$2 \times 10^{-7} \text{ cm}^2/\text{s}$	

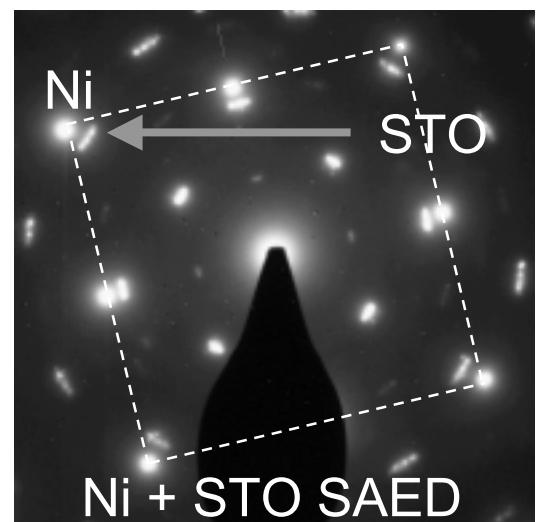
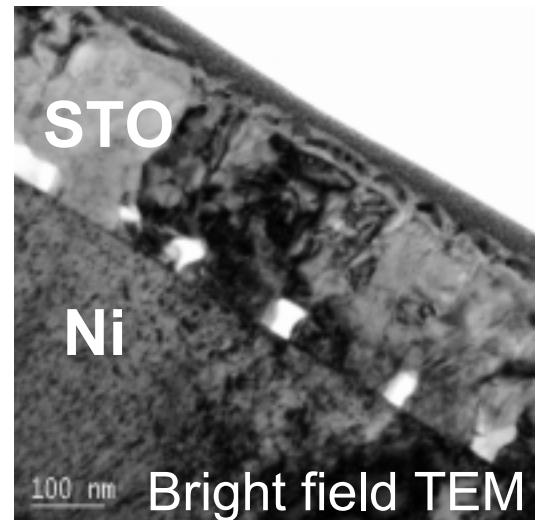
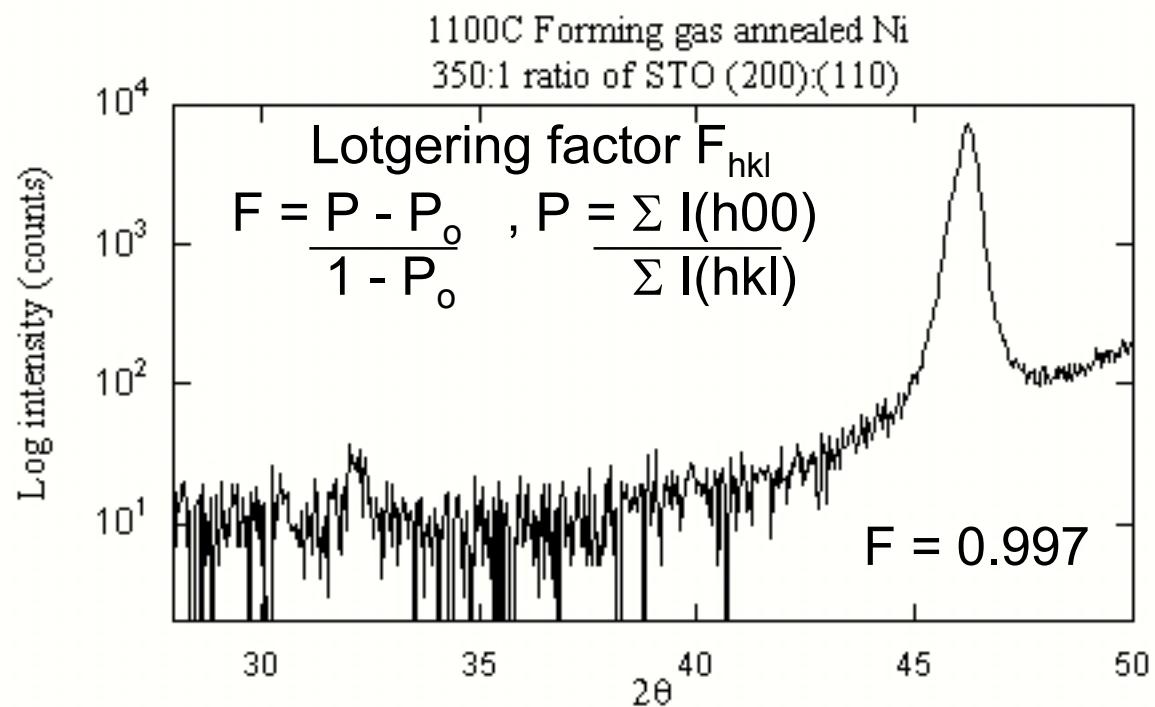


Solution-deposited epitaxial SrTiO₃ on Ni

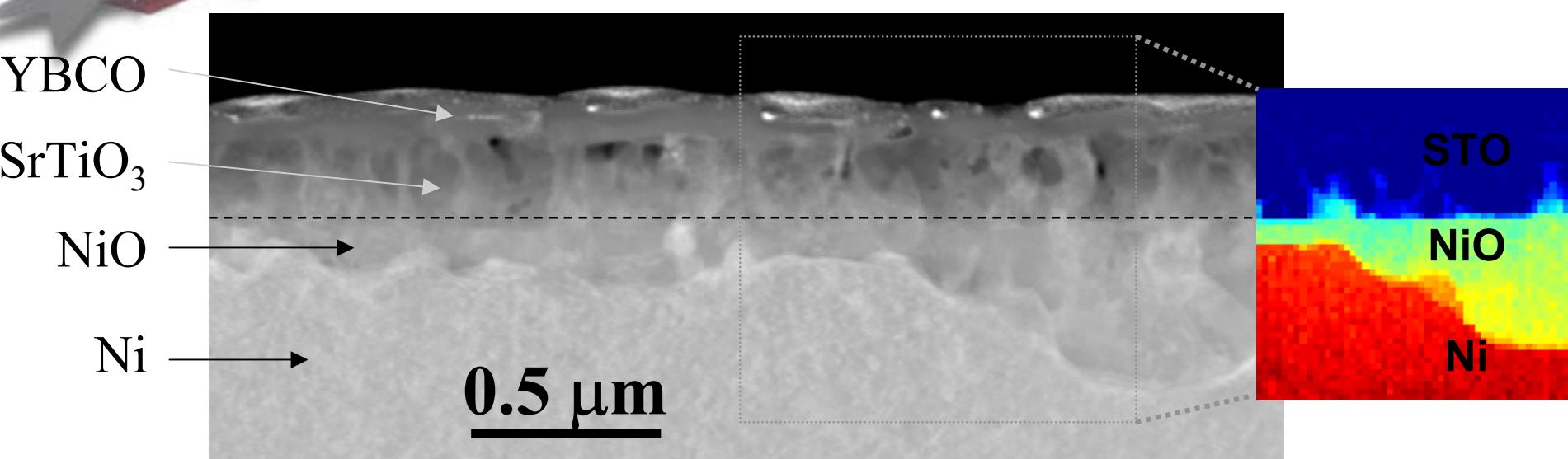
TFA addition avoids carbonates in sol-gel YBCO and other oxides, including BaTiO₃ and SrTiO₃.

Enables epitaxial SrTiO₃ growth as low as 500°C.

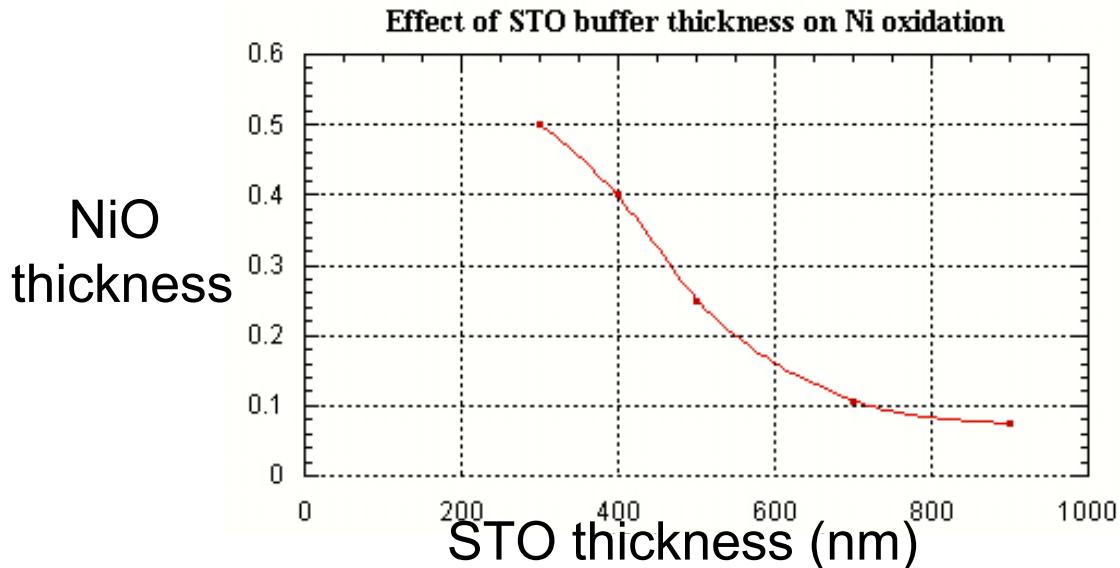
Currently depositing SrTiO₃ on Ni/Ni-W



Control of Ni substrate oxidation



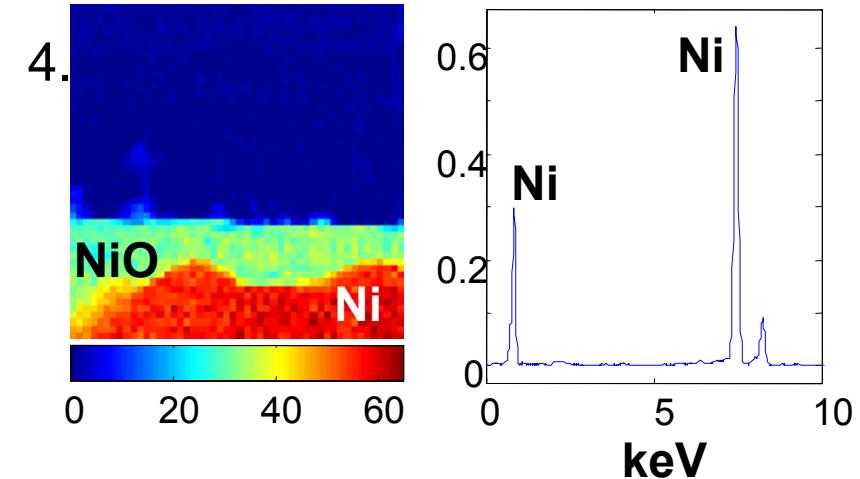
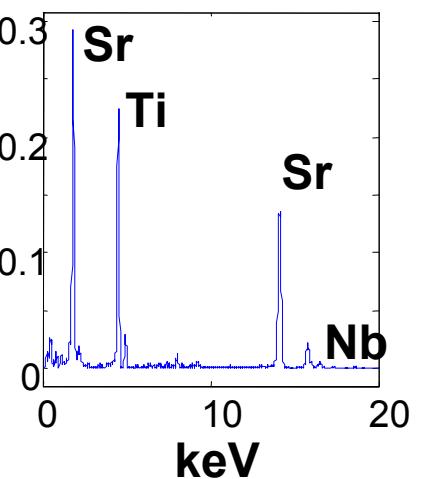
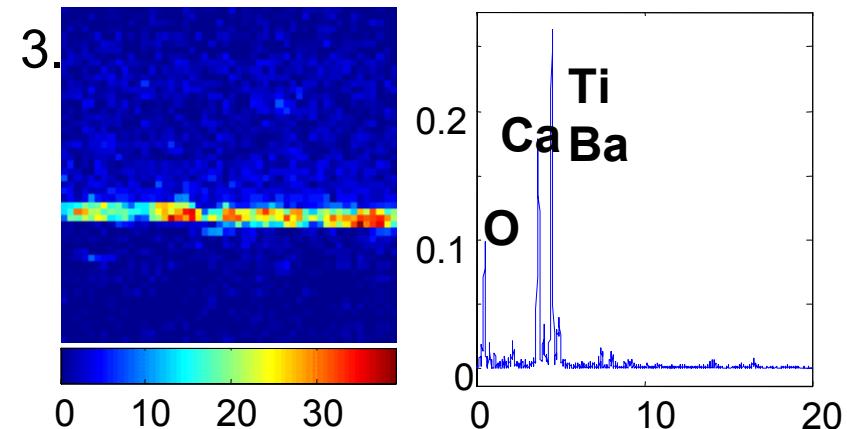
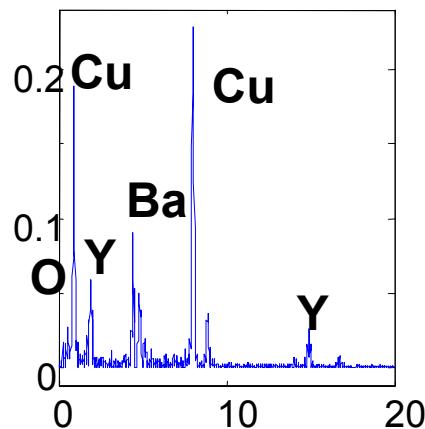
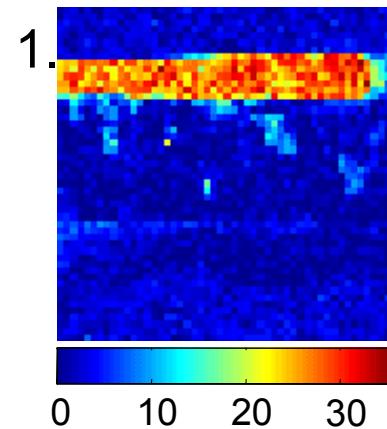
Increasing SrTiO₃ thickness suppresses Ni oxidation.
STO \geq 500nm has been the best support for YBCO to date.



Results for YBCO
grown on STO/Ni
at 800°C, 30 min

Process Integration TEM: X-ray Spectral Image Analysis

YBCO/STO/BCT/Ni(100)





2003 Development Goals

Increase density in buffer layer (suppress NiO formation)

Scale-up from 25 cm to > 1 meter lengths with ORNL ACCI facility

Collaborate with ORNL on TFA-YBCO conversion

- ultimate goal: 1 μm YBCO, $J_c = 2\text{MA/cm}^2$, 200 A/cm

TFA-YBCO conversion time/process is key limitation

Production/cost predictions

Capacity: coating at 3-10 cm/s, 1-3 million meters/year

Easy regeneration: 1 km YBCO \approx 100 ml of solution

YBCO, buffer materials cost (today, 500g lots) < \$ 1/kA-m

Expect: 0.25 μm YBCO, 2MA/cm² on buffer/Ni \rightarrow 50A/cm

1.0 μm YBCO, 1MA/cm² on buffer/Ni \rightarrow 100A/cm



Acknowledgments

Oxford Superconducting Technologies

- Ken Marken

Oak Ridge National Laboratories

- D. Lee, M. Paranthaman, D. Christen, R. Feenstra, A. Goyal

American Superconductor Corporation

- Martin Rupich

This work was supported by the US Department of Energy
DEER/Superconductivity Program for Electric Systems.

Sandia National Laboratories is a multiprogram laboratory
operated by Sandia Corp., a Lockheed Martin Company, for the
US Department of Energy under contract DE-AC04-94AL85000.